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# DISMANTLING STRATEGIES FOR HIGHLY HCH-POLLUTED LANDFILL LEACHATE POND USING ELECTROCHEMICALLY ASSITED TECHNOLOGY

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# Introduction







Old landfill leachate pond:

■ HPDE sheet and dimension of 50x25 m<sup>2</sup>

 Thickness of more than 1 m of sludge (saturated and without compacting)

Heterogeneous HCH contamination.
 From few mg/kg to dozen g/kg

It is currently collecting runoff and leachate drained in the ditch and leachate pumped from boreholes, which includes contributions of DNAPL microdroplets







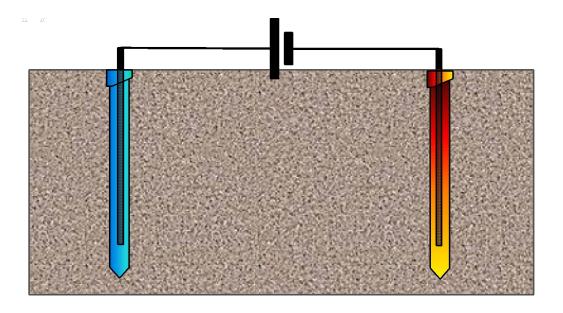
# Sludge characteristics:

- **✓** black color, with gray tones on the surface by oxidation
- ✓ very plastic, adherent and difficult to handle.



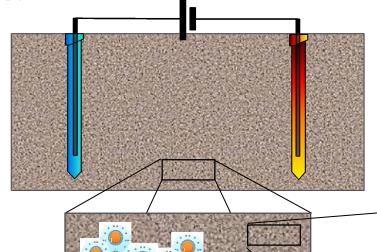


# What happens when we apply an electric field to electrodes placed in the sludge?



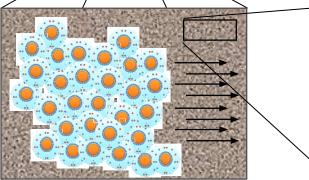
Includes *different processes* that take part in a soil:

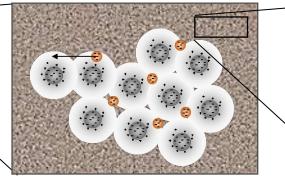
- Physical processes (heating)
- ☐ Chemical processes (ionic exchange, precipitations, etc.)
- ☐ Electrochemical processes (water oxidation and reduction)
- ☐ Electrokinetic processes (electrosmosis, electrophoresis and electromigration)

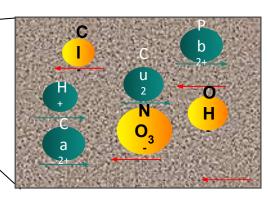


## **Electrokinetic processes**

Includes *different processes* that take part in a soil as a result of the application of an electric field.







#### **MACROSCOPIC LEVEL**

# Electro-osmosis movement of the water into the soil through the application of electric field.

MICROSCOPIC LEVEL

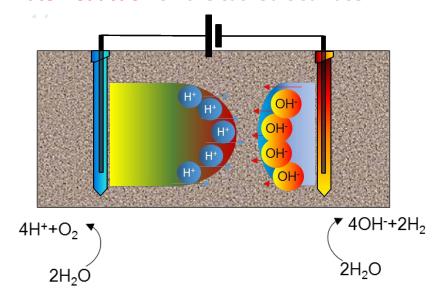
# Electrophoresis movement of the charged particles into the soil through the application of electric field.

#### **MOLECULAR LEVEL**

# Electromigration movement of the ions into the soil through the application of electric field.

#### **Electrolysis**

Water oxidation on the anodic surface Water reduction on the cathodic surface

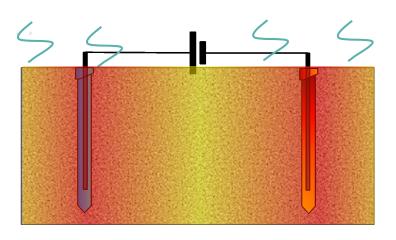


#### Influences on:

Different species precipitation / redisolution
 Fixation or liberation of species by ionic exchange
 Microbial viability

#### Soil electrical heating

Temperature increase that occurs in the soil due to the ohmic drop generated by the high electrical resistance of the soil.



#### Influences on:

- Viscosity changes in liquids contained in the soil
- Pollutants volatilization
- Water evaporation



# Objectives

# TO DISMANTLE HIGHLY HCH-POLLUTED LANDFILL LEACHATE POND USING ELECTROCHEMICALLY ASSITED TECHNOLOGY

The ideal conditions for dismantling the pond would be:

- To reach a level of dryness in the sludge that would allow its mechanical handling without risk.
- To minimize the atmospheric emissions.
- To minimize the need for necessary infrastructure and equipment, as well as subsequent treatments for conditioning, transfer, or decontamination

# **Partial technical objectives**

To determine the mobility of pollutants in the sludge under the action of electric field.

To evaluate the dewatering of the sludge by the action of the electric field.

To study the
treatment of
water
mobilized and
recovered from
the treatment by
electrochemical
technologies.

To evaluate the implementation of GAC column to retain organics transfer to the atmosphere

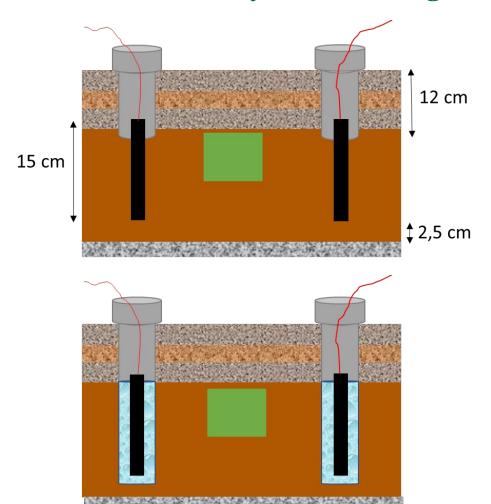
To implement the technology in real scale

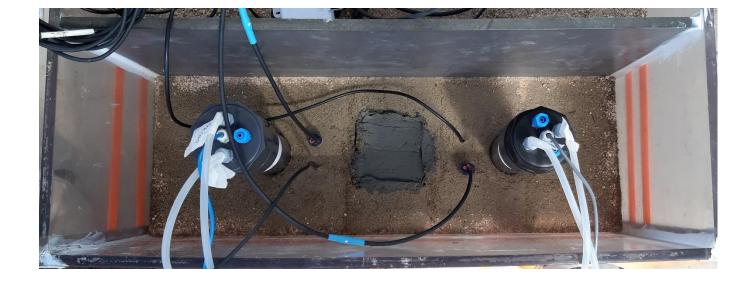


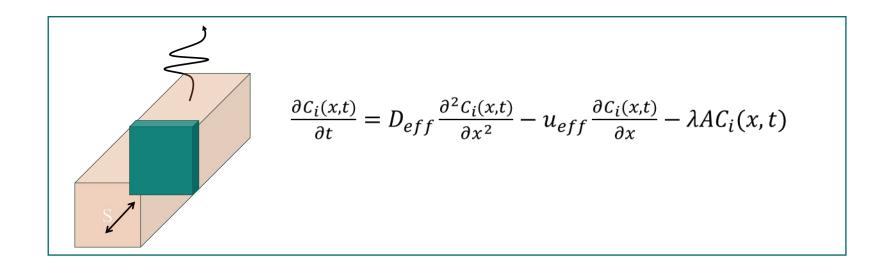
# Results

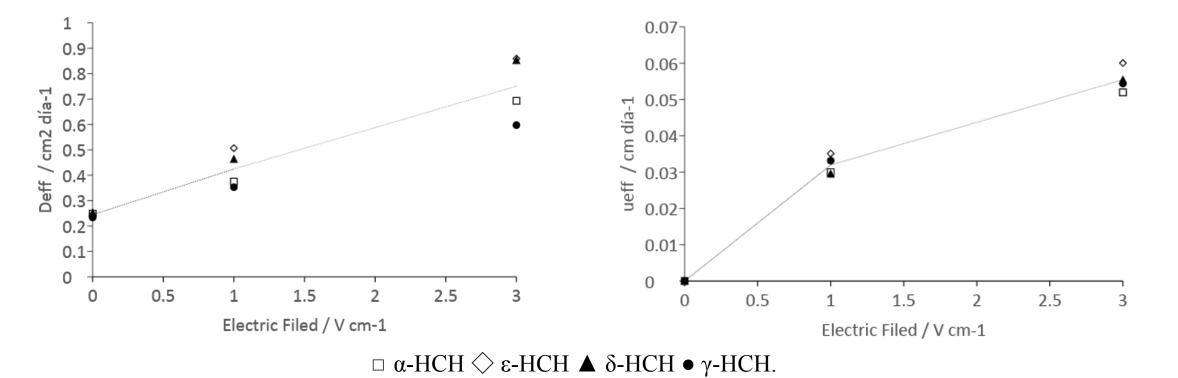
 
 ▼ Field
 Electrochemical Study of **EK treatment** mobilization and dewatering treatment of application of the sludge at liquid streams bench scale

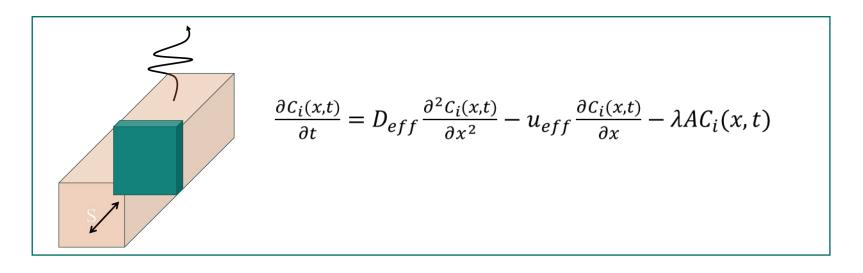
Characterization of sludge water and contaminant mobility under the action of the electric field during the first electrochemically assisted stage. Stage 1.

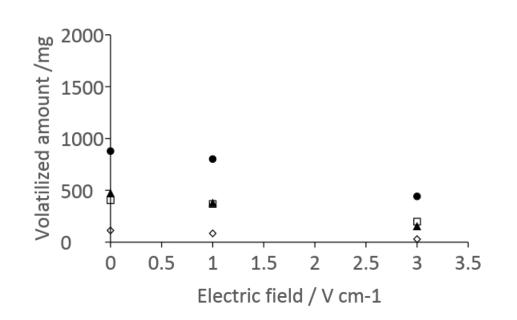


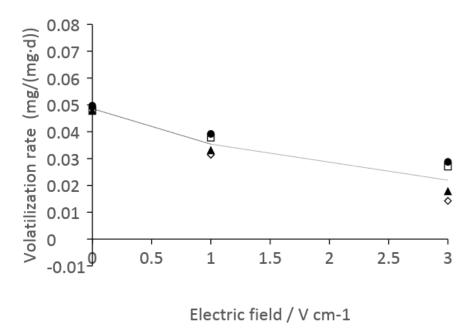








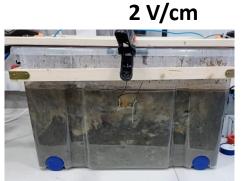


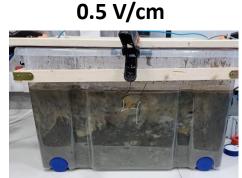


 $\Box$   $\alpha$ -HCH  $\diamondsuit$   $\epsilon$ -HCH  $\blacktriangle$   $\delta$ -HCH  $\bullet$   $\gamma$ -HCH.

Characterization of sludge water and contaminant mobility under the action of the electric field during the first electrochemically assisted stage. Stage 2.







50 kg sludge / test

1 V/cm + SDS

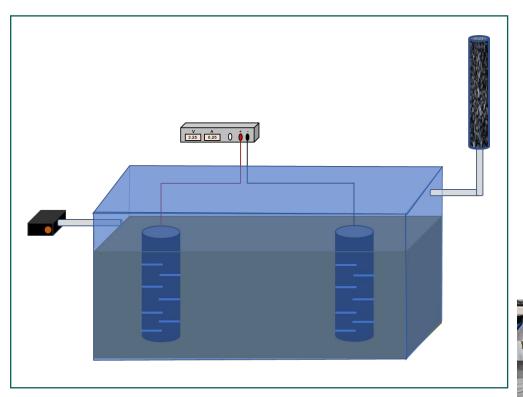


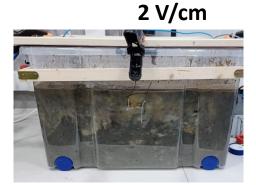


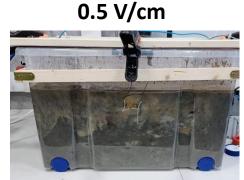
1 V/cm + iron electrodes



Characterization of sludge water and contaminant mobility under the action of the electric field during the first electrochemically assisted stage. Stage 2.

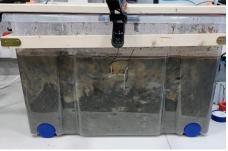






50 kg sludge / test

1 V/cm



1 V/cm + SDS

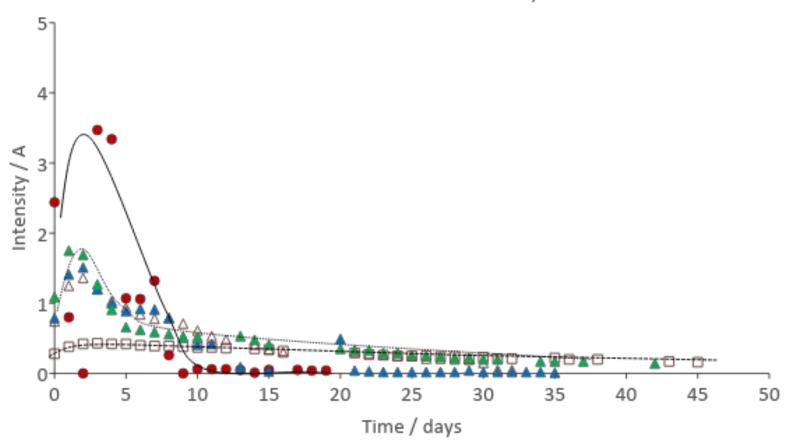


1 V/cm + iron electrodes

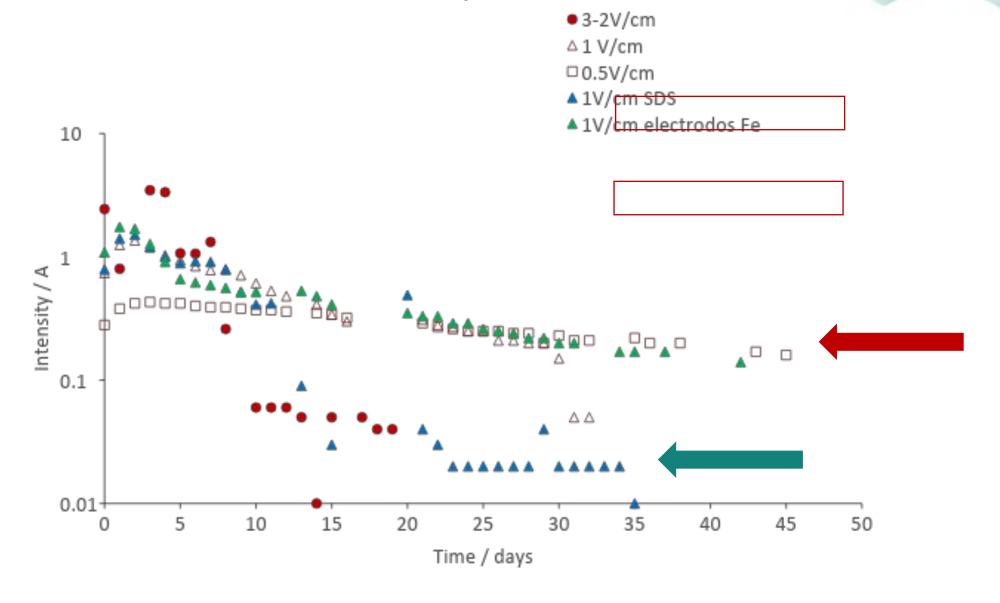


### General behaviour. Current intensity.

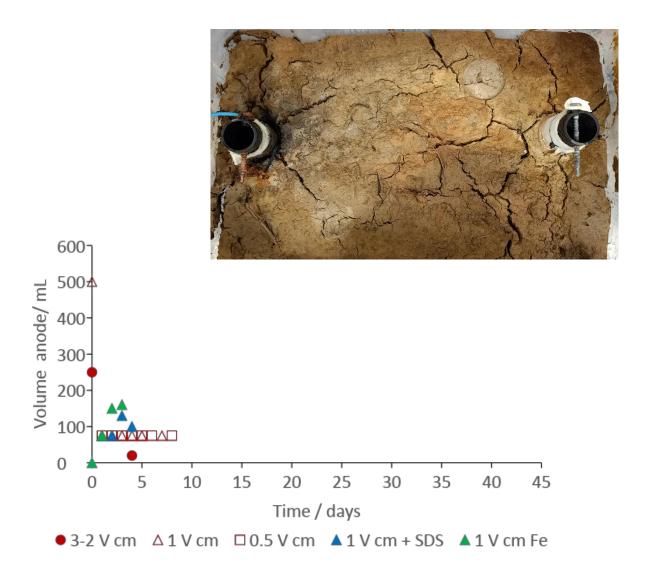
- 3-2V/cm
- △1 V/cm
- □ 0.5V/cm
- ▲ 1V/cm SDS
- ▲ 1V/cm electrodos Fe

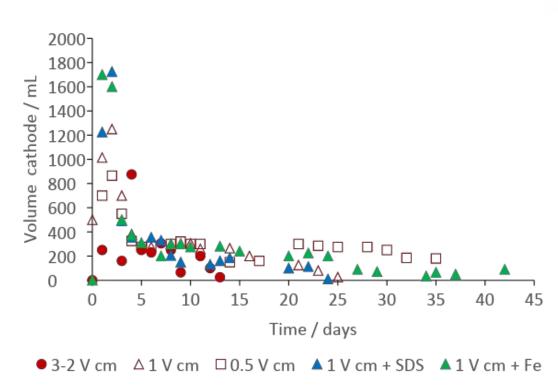


### General behaviour. Current intensity.

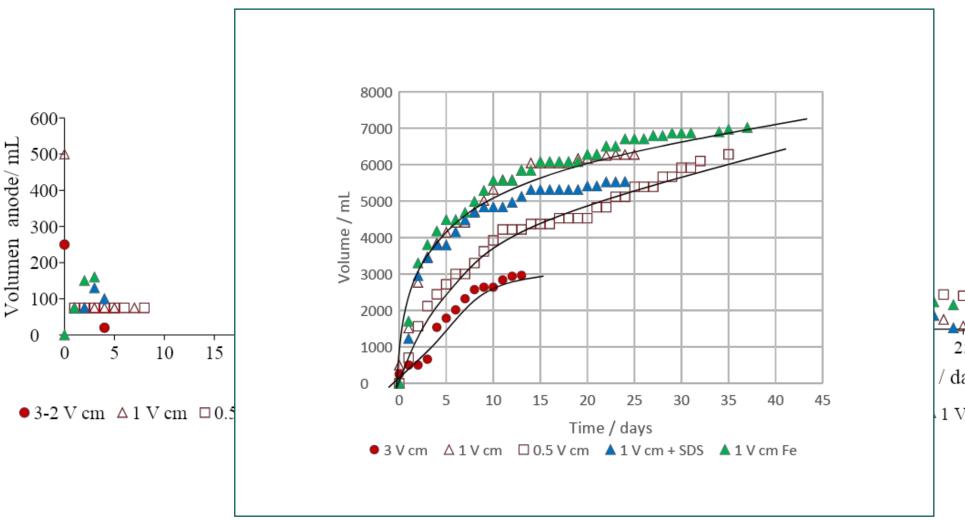


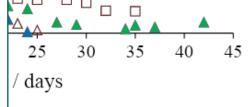
#### General behaviour. Transport of water.





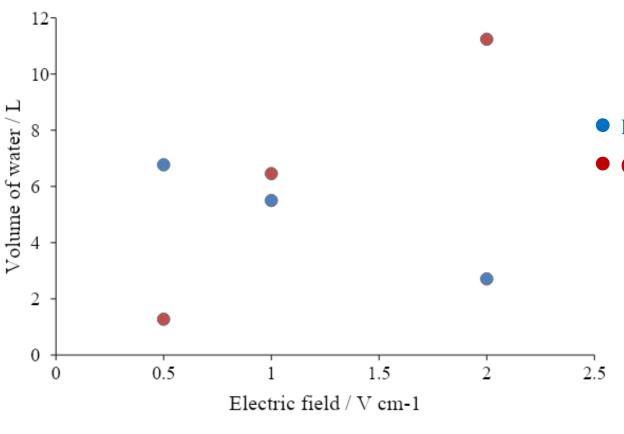
### General behaviour. Transport of water.





1 V cm + SDS ▲ 1 V cm + Fe

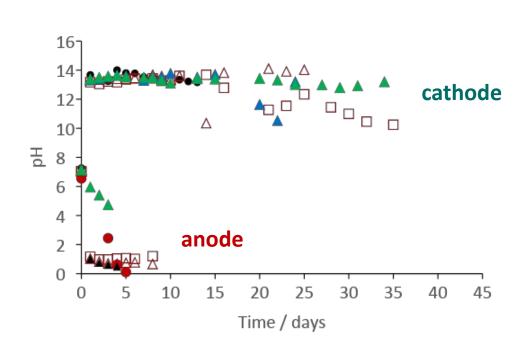
## General behaviour. Transport of water.

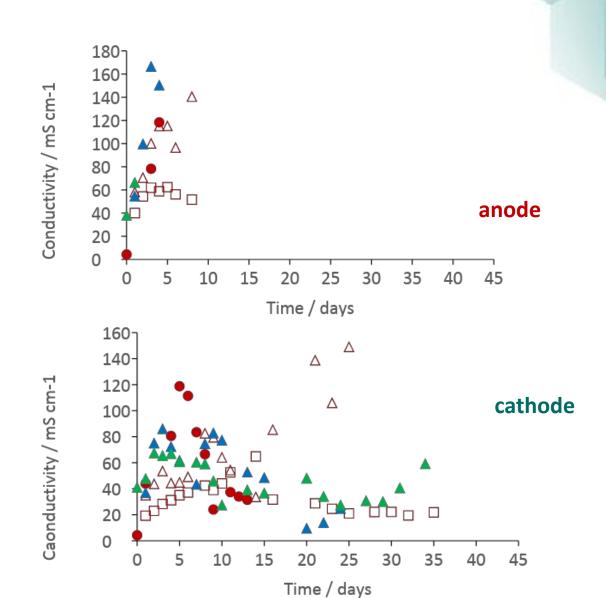


- mobilized by EK
- evaporated

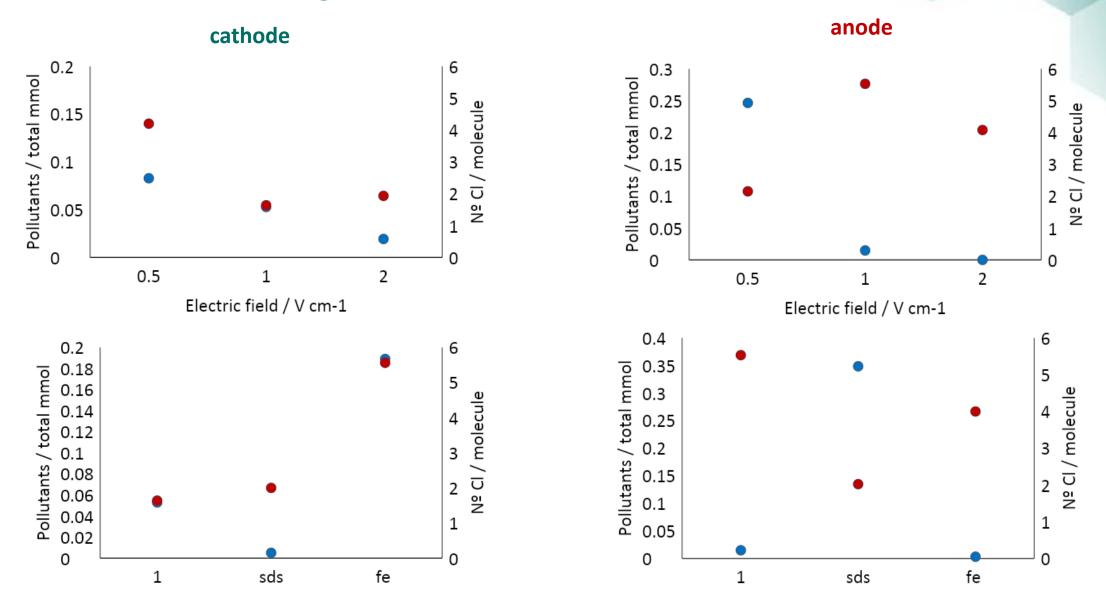


#### General behaviour. pH and conductivity of mobilized water.



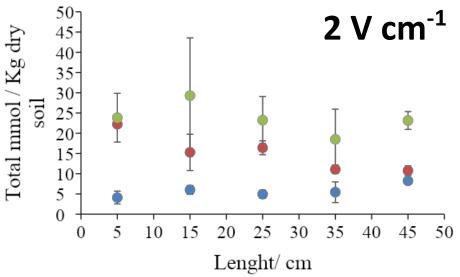


### General behaviour. Organics mobilized with water.

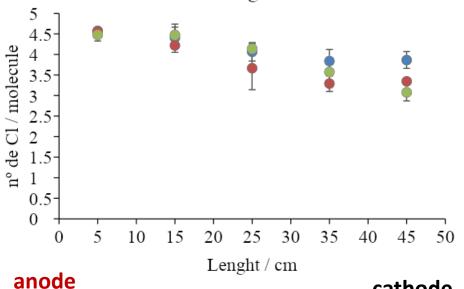


### General behaviour. Organics in treated sludge.

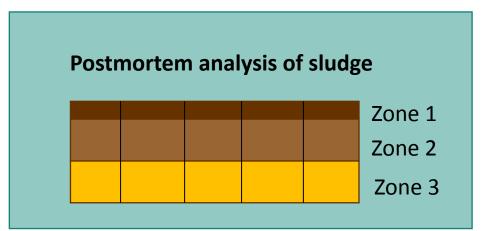
cathode



Average zone 1= 5.75 mmol / Kg SS Average zone 2= 15.16 mmol / Kg SS Average zone 3= 23.61 mmol / Kg SS

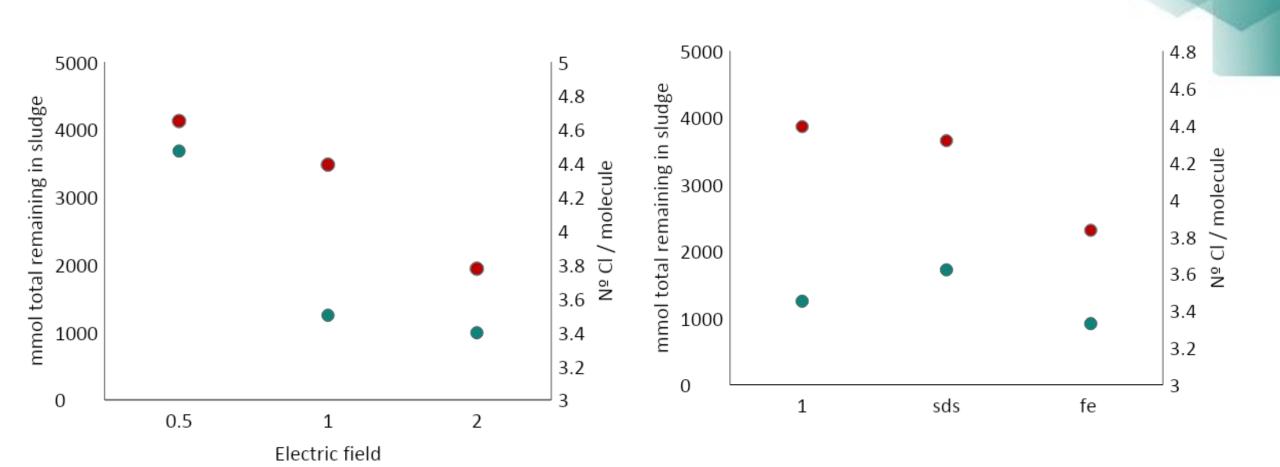


Initial average value = 4.68 Final average value=3.78



#### General behaviour. Organics in treated sludge.

#### Postmortem analysis of sludge

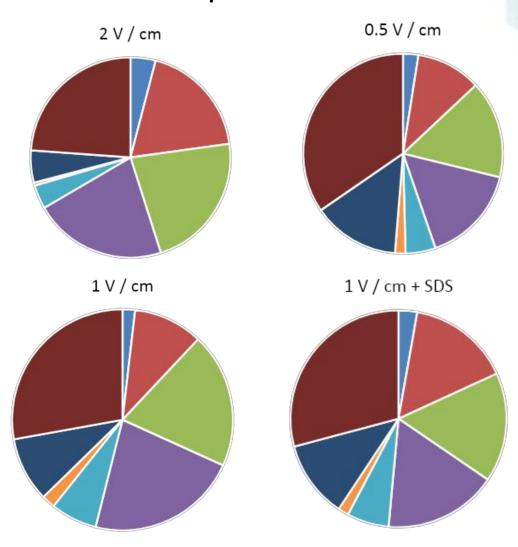


### General behaviour. Organics in treated sludge.

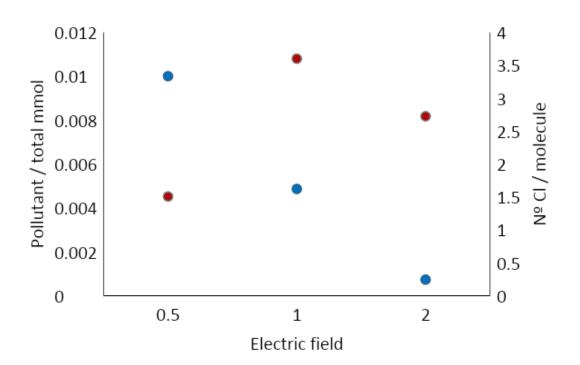
#### **Initial pollutant distribution**

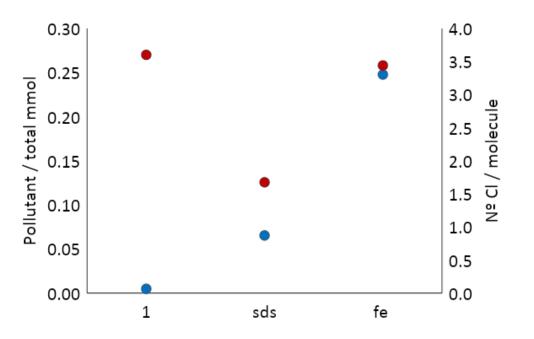
# CB ■suma DCB ■suma TCBs ■suma Tetra CB suma PentaCB + PentaCX ■suma HexaCX ■suma HeptaCH HCHs

#### Final pollutant distribution



### General behaviour. Organics retained in activated carbon.





### **Expected results**

# **Conclusions**

Draging of pollutants with electroosmotic flux

Pollutants are strongly retained in sludge. Low mobilization

Mobilization of water by EK

Mobilization of water by EK, but high cell voltaje leads to lower EO fluxes

Evaporation of water favoured at high cell voltage

Evaporation of water is favoured at high cell voltaje

SDS helps to drag HCH

Negative effect of SDS

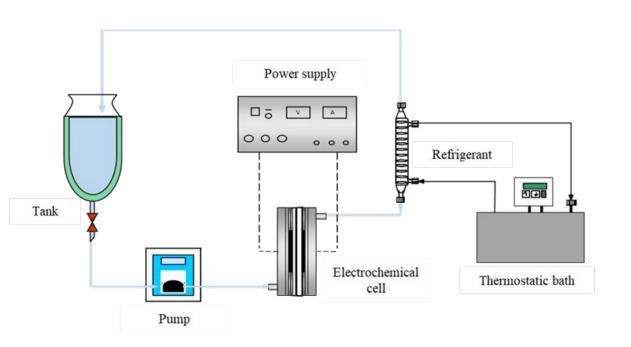
Iron electrodes has positive effect

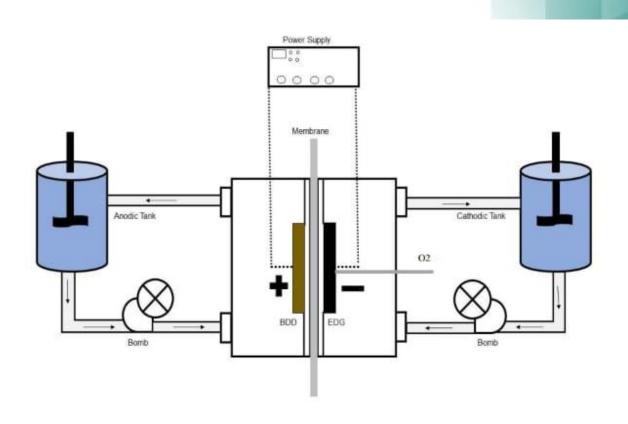
Iron electrodes contirbute to minizime acid front and dehalogenation

COVs can be transfered to gas phase

COVs are stripped and they can be retained in GAC, but water trap should be implemented

#### Electrochemical treatment of extracted water.





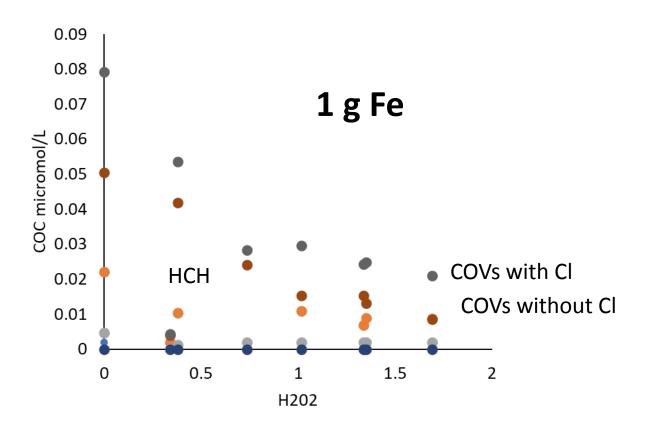
Electrolysis PhotoElectrolysis

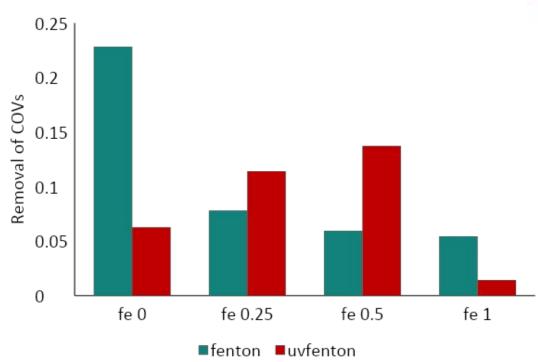
Electrodes: BDD and MMO

Electrofenton
Photoelectrofenton
H2O2/Fe ratio

#### Electrochemical treatment of extracted water.



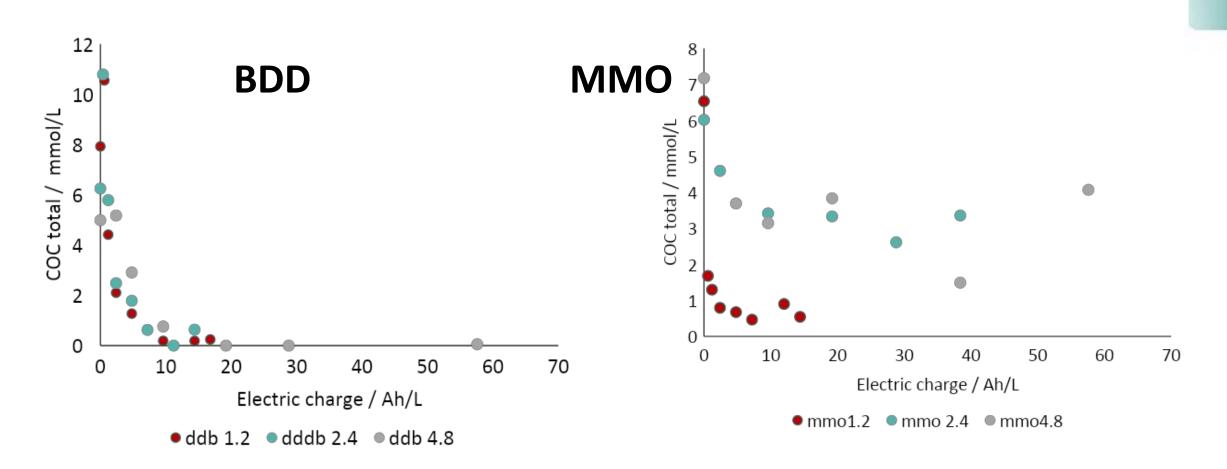


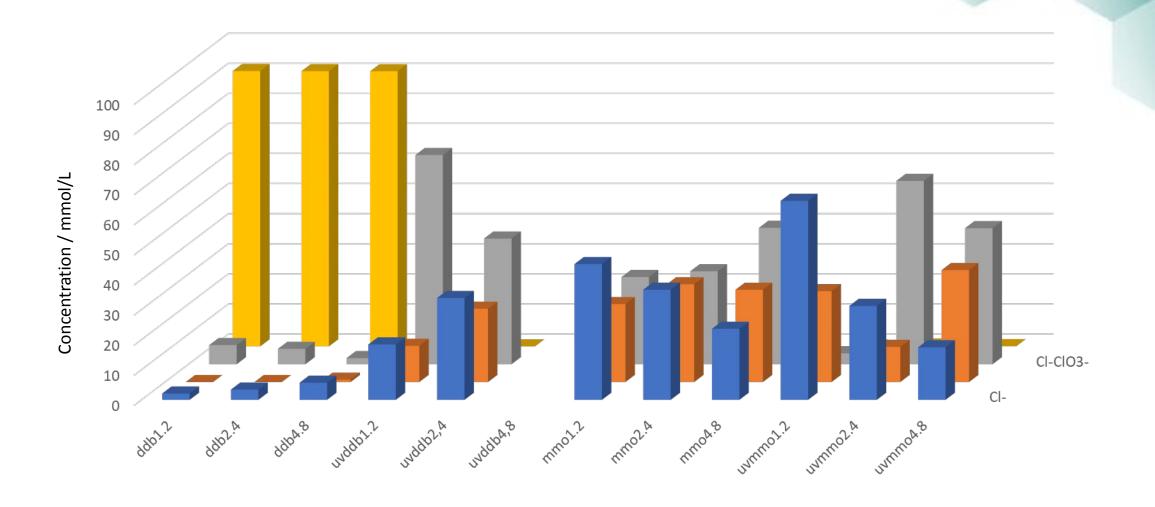


#### Electrochemical treatment of extracted water.

#### Electrolysis PhotoElectrolysis

Electrodes: BDD and MMO





## **Conclusions**

Low efficiency of E-fenton Process.

H2O2 can partially degrade HCH

Electrooxidation with BDD is able to attain complete removal of HCH and derivates, but perchlorate is also generated

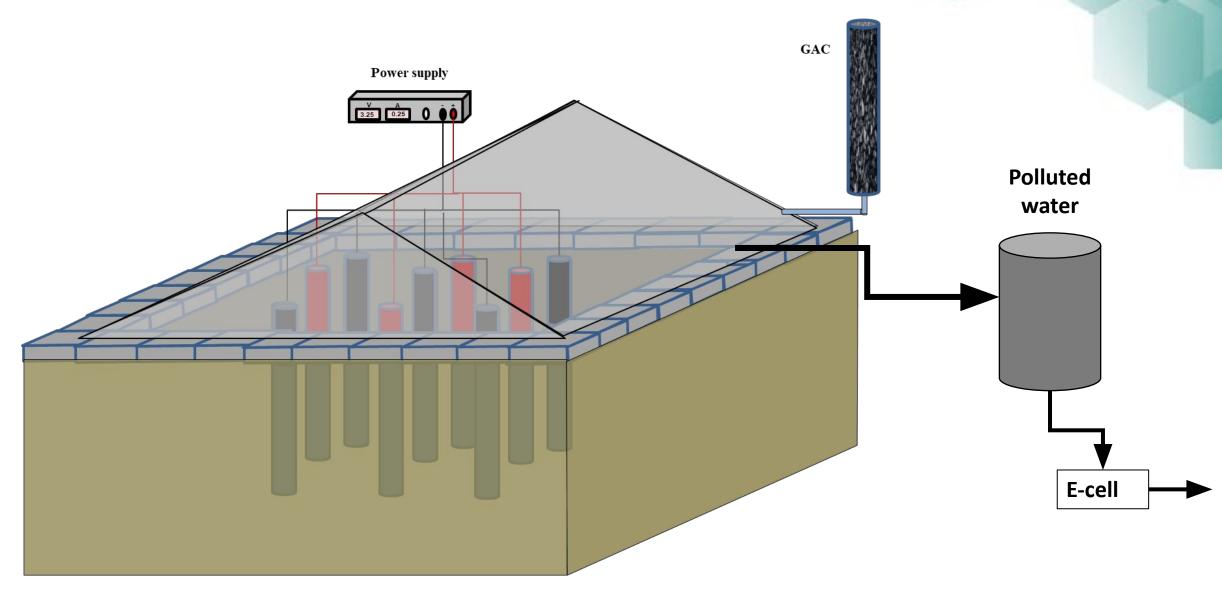
Low efficiency of MMO electrodes

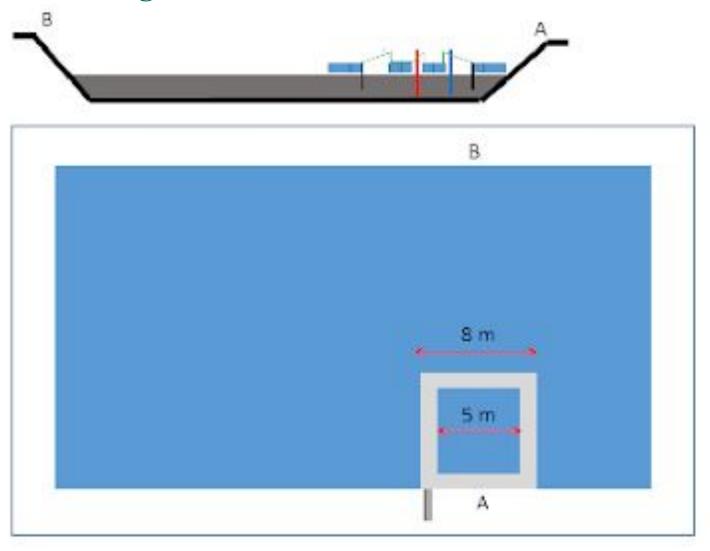
UV irradiation negatively affects COVs degradation

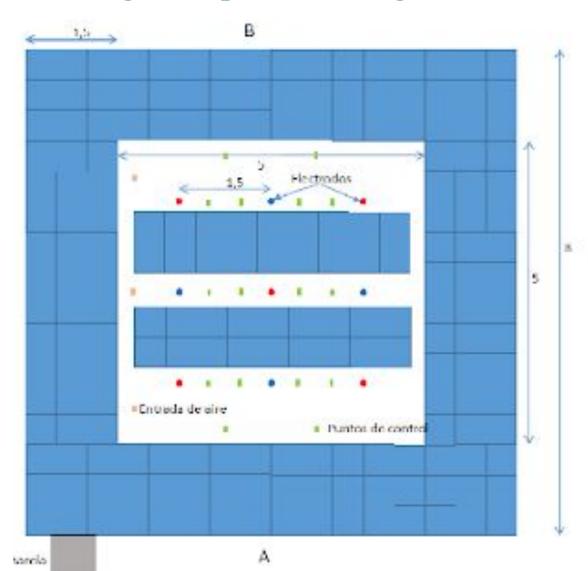
Direct process instead of radicalary process



Dismantling sludge dump



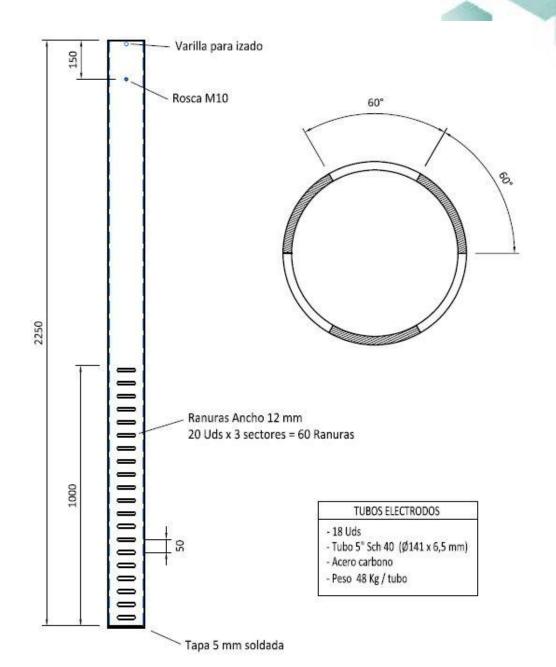




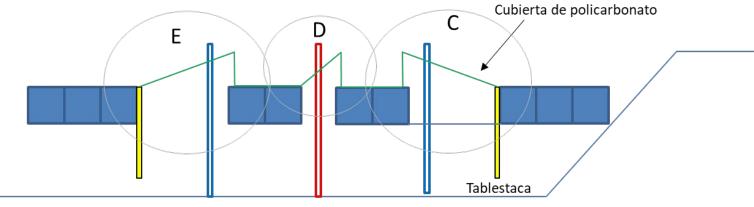


#### **Electrodes:**

Iron hollow and perforated tube



#### **Cover:**

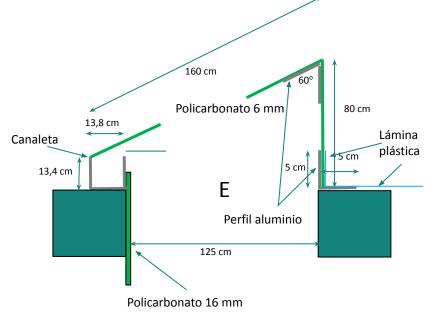


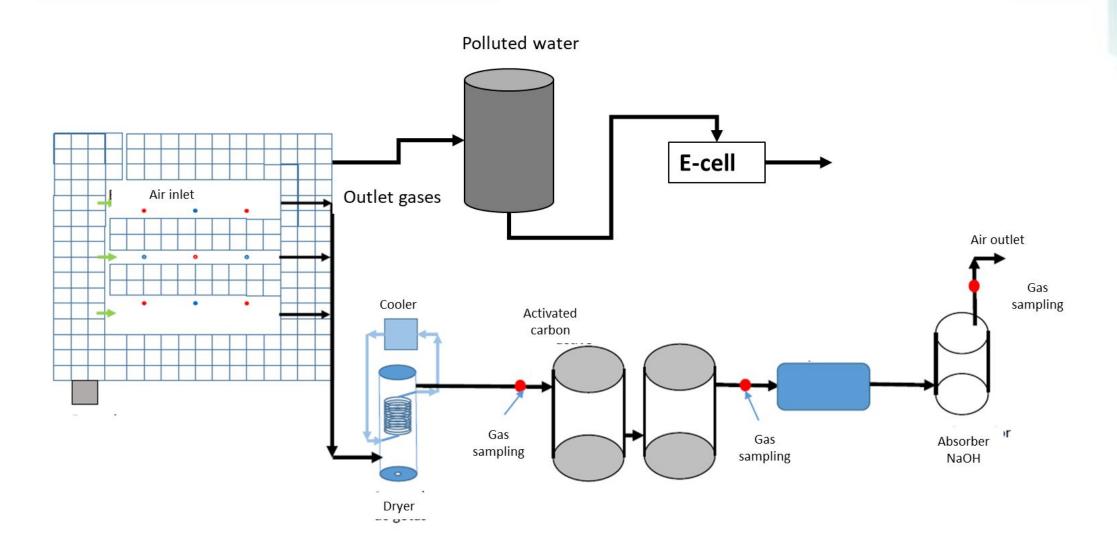


Polycarbonate plate ( to cover)



Water collection channel











### THANK YOU FOR YOUR ATTENTION

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